

Determination of Nitrofurans Metabolites in Milk by Liquid Chromatography with Diode-Array Detector

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Abstract

For the analysis of four metabolites of nitrofurans in raw milk - furazolidone, furalfadone, nitrofurazone and nitrofurantoin - a liquid chromatographic method using a diode detector was used. The method complies with the requirements of European Commission Resolution 2002/657 / EC. The sample was extracted by ethyl acetate, liquid-liquid extraction method was used, cleaning done by solid phase extraction on a silica gel column, after sample hydrolysis and derivatization with 2-nitrobenzaldehyde. The validation of the method was conducted following the European Union criteria for the analysis of veterinary drug residues in foods. The decision limits (CC α) were 0.14-0.32 $\mu\text{g}/\text{kg}$, and the detection capabilities (CC β) 0.18-0.39 $\mu\text{g}/\text{kg}$. The advantage of the method is that with relatively less financial costs it is possible to determine less than the minimum working limit for nitrofurans metabolites set by the EU (MRPL Minimum Required Performance Limit 1 $\mu\text{g}/\text{kg}$). This method is financially acceptable for developing countries.

Keywords: LC/DAD; Nitrofurans metabolites; Furazolidone; Furalfadone; Nitrofurazone; Nitrofurantoin; Milk

Highlights: Food safety requirements. Determination of nitrofurans in milk. Developing cheaper alternative for determination of nitrofurans residues. Developing inexpensive method for developing countries.

Introduction

Nitrofurans belong to the group of broad-spectrum synthetic antibiotics which are effectively used in veterinary medicine for the prevention and treatment of such gastrointestinal infections such as *Escherichia coli*, *Salmonella spp.*, *Mycoplasma spp.*, *Coccidia spp.*, Coliforms and protozoa which are found in animal products and water [1-3]. These medicines are rapidly metabolized within a few hours after ingestion, and residual metabolites can remain in the organism for several weeks and possibly months in the form of connected protein [4,5]. It was confirmed that these metabolites of

nitrofurans represent a potential risk to human health due to their carcinogenic, teratogenic and mutagenic effects [6,7]. Due to food safety requirements, EU restricts the use of nitrofurans preparations in veterinary medicine due to their toxic, carcinogenic and mutagenic properties. [8,9]. After the EU the use of nitrofurans in veterinary medicine has been restricted in other countries too, in the United States, China, and Japan [4,10] However, due to their low cost and significant effectiveness, the use of nitrofurans is allowed or used illegally as a veterinary drug in some developing countries [11,12]. As of today, in the EU, in poultry and seafood, marginal permissible

norm of the mentioned four nitrofurans (MRPL - Minimum Required Performance Regulation 1442/95) is 1 µg / kg [1,13-15]. Illegal use of nitrofurans in the EU is controlled by official inspection and analysis services in accordance with the requirements of Council of Europe Directive 96/23 / EC. In the interest of exports, third world countries are forced to adopt the MRPL L set by the Council of Europe and thus reach the same threshold as EU laboratories [16]. Given the strict regulations and validation requirements of analytical methods set by the Council of Europe, the development of highly sensitive and specific analysis methods for the determination of nitrofurans residues in foodstuff is becoming an increasingly difficult task. (USP 2022; Guidance for Industry Bioanalytical Method Validation 2001; ICH Q2 A [17]. In accordance with the requirements of the European Union, the use of nitrofurans in stockbreeding is regulated by the legislation of Georgia, in particular, by the Resolution №499 of the Government of Georgia, the technical regulation is approved. Rule for the implementation of methods of analysis and interpretation of results for the examination of certain substances (substances) in animal and food of animal origin and their waste [18] which places quite strict requirements on the analytical methods used for this purpose. As of today, the analytical strategy for the quantification of nitrofurans is based on the determination of 4 stable and steady metabolites which can be released from proteins in the weak acid solution and then derivatized [19,20]. These stable metabolites are 3-amino-2-oxazolidone AOZ, 3-aminomorpholinomethyl-2-oxazolidinone AMOZ, 1-aminohydantoin AHD and semi carbazide SEM [21]. Many analytical methods for the determination of nitrofurans metabolites have been developed in various matrices, such as seafood [22,23], animal feed [24], meat [25], milk [26], honey [27-29] and others [30,31]. These analytical methods are liquid chromatography-tandem mass spectrometry (LC-MS), [32] high-performance liquid chromatography-tandem mass spectrometry (HPLC-MS/MS), [33] ultra-high-performance liquid chromatography-tandem mass spectrometry (UPLC-MS/MS), [34,35] enzyme-linked immunosorbent assay (ELISA), [36], Ultrapformance Liquid Chromatography-Photodiode Array Detection [37] etc. [38,39]. Not all of these methods meet the requirements due to the very low detection threshold required for the study substance MRPL = 1,0 mkg/kg). [40] Therefore, for its high sensitivity and accuracy of HPLC-MS / MS [41]. is mainly used the use of this method of analysis is limited due to the hardware and the high cost of servicing this hardware [31] Which is why it is still not possible to introduce this method in economically developing countries [42]. Therefore, it is relevant to develop a relatively inexpensive and simple analytical method for the determination of nitrofurans metabolites in food of animal origin. Liquid chromatographic method using diode detector is relatively less expensive and convenient compared to other detection methods listed above however, there is almost no data on this which is due to the fact that it is difficult to achieve the required detection limit using diode detector. [39,40,43]. In order to simplify the experimental process according to the requirements of MRPL, after extraction, we used the solid-phase cleaning to reach required detection limit. For solid-phase cleaning, instead of expensive cartridges (columns), we used our own silica gel columns. We have selected milk as a research matrix, as milk control ensures the

safety of all dairy products. As a result, we have obtained a sensitive, fast and relatively inexpensive method that meets the requirements of a set MRPL. The method can be successfully used for the simultaneous determination of four metabolites of nitrofurans in milk and guarantees the safe use of milk and consequently products made from this milk. This method is financially affordable and can be used in developing countries.

Materials and Methods

Reagents and Chemicals: Hexane (95%), anhydrous dipotassium hydrogen orthophosphate, 35% hydrochloric acid, sodium hydroxide, 2-nitrobenzaldehyde (2-NBA), acetonitrile (HPLC grade), dimethyl sulfoxide anhydrous, ≥99.9%, ethyl acetate, sodium acetate ACS reagent, ≥99.0% were from Sigma Aldrich Chemical Company (Germany) and Kieselgel 60 from Roth (Germany). Ultrapure water was filtered through a Milli-Q system Millipore (USA). The metabolites AOZ, AMOZ, AHD, and SEM, and the internal standards AOZ-d4, AMOZ-d5, (2-NP-¹³C₃) AHD and 1,2- (¹³C¹⁵ N₂) SEM were supplied by Sigma (Aldrich Chemical Company, Germany).

Standard solutions

Individual standard stock solutions of 1 mg/mL were prepared in acetonitrile. Working solutions of 10 ng/mL were diluted by water. All standard stock solutions were stored -20°C, and the working solutions were stored in refrigerator. The concentration and content of mix standard solution was used to spiked samples with AMOZ, AOZ, AHD and SEM at a 8, 16, 20, 28 and 36 ng/mL respectively. The concentration and content of internal mix standard solutions were used AOZ-d4, AMOZ-d5, (2-NP-¹³C₃) AHD and 1,2- (¹³C¹⁵ N₂) SEM at a 40, 40, 100 and 100 ng/mL, respectively.

Preparation of silica gel columns

The silica gel column was prepared by the dry method (18 x 75 mm). The prepared column was washed with 5 ml of hexane, dried under vacuum, then washed with 10 ml of acetonitrile, and again dried under vacuum.

Sample preparation

Nitrofurans and their metabolites were prepared using the same method as previously described, with some modifications [44,45]. The cooled milk was centrifuged at 3500 rpm for 15 minutes at +4°C and the fat layer was carefully removed. 2.0 ± 0.05 g of homogenized raw milk was weighed into a 50 ml polypropylene centrifuge tube. Standard spiking solution mix (50, 100, 150 and 200 µL), internal standard solution mix (100 µL), 5 mL of 0.1 M hydrochloric acid solution and 50µl of a solution of 2-nitrobenzaldehyde in dimethyl sulfoxide (DMSO) (8 mg ± 0.6 mg in 5 ml of DMSO) were added. Thoroughly stirred for 1minute (with vigorous stirring, the extract acquires a jelly-like consistency) and incubated overnight at 37°C to hydrolyze protein-bound NP metabolites and convert the metabolites to their nitrophenyl (NP) derivatives. After the sample solution was cooled to room temperature, 500 µl of di-potassium hydrogen orthophosphate solution, at least 300 µl of sodium hydroxide solution were added to adjust the pH to 7.0 ±

0.5, and 5 ml of ethyl acetate; Thoroughly mixed for 1 minute (with intensive stirring, the extract acquires a jelly consistency); centrifuged at 3500 rpm for 15 minutes at +4°C, ethyl acetate layer transferred to a silica gel column for sample purification at a rate of 5 ml / min. After completion of sampling, the column was washed with 10 ml of ethyl acetate, the column pumped out with a vacuum, the eluates were combined and was evaporated in a moderate flow of nitrogen, the residue was dissolved in 1 ml of hexane, 1 ml of the mobile phase was added, mixed for 1 minute, centrifuged at 3500 rpm for 15 minutes at +4°C, the hexane layer was removed and the resulting aqueous phase was used for chromatography.

Instrumentation

1.1.1.1. **LC/DAD:** The LC/DAD system consisted of an Agilent Series 1260 HPLC system (Agilent Technologies, Germany) with DAD detector. The chromatography was performed in a C18 column 3 μm x 2 mm 150 mm (Phenomenex, Torrance, CA, USA), connected to a C18 precolumn 3 μm x 2 mm x 4 mm (Phenomenex, USA). The mobile phase was Acetonitrile: 0.01 M sodium acetate buffer pH 6, 0 - 250:750, λ 376 nm, flow rate of 1 mL/min, Injection volume was 50 μL . The column was thermostated at 30°C. All determinations

were carried out under standard conditions: Air temperature - (20 ± 5) °C, atmospheric pressure - 84.0 - 106.7 kPa (630 - 800 mm Hg), air humidity no more than - 80%, mains voltage 198 - 242 V, frequency AC - 50 ± 1 Hz.

Results

Specificity/Selectivity

Specificity/selectivity were evaluated via analysis of blank matrix samples fortified with standards of nitrofurans metabolites (concentration of 1 $\mu\text{g}/\text{kg}$ each). According to analysis for the studied substances of the nitrofurans group, the specific wavelength is 376 nm (Figure 1), no significant peaks with an S/N (signal to noise) ratios of 3 or more and chromatographic interference were being observed at the retention times of the targeted nitrofurans metabolites (Figure 2); The coefficient of variation of the specificity of the results obtained during the working day is in the range of 0.02-0.17, during the working week - in the range of 0.02-0.26. Which indicates satisfactory, required by Decision 2002/657/EC (Figure 1 and 2).

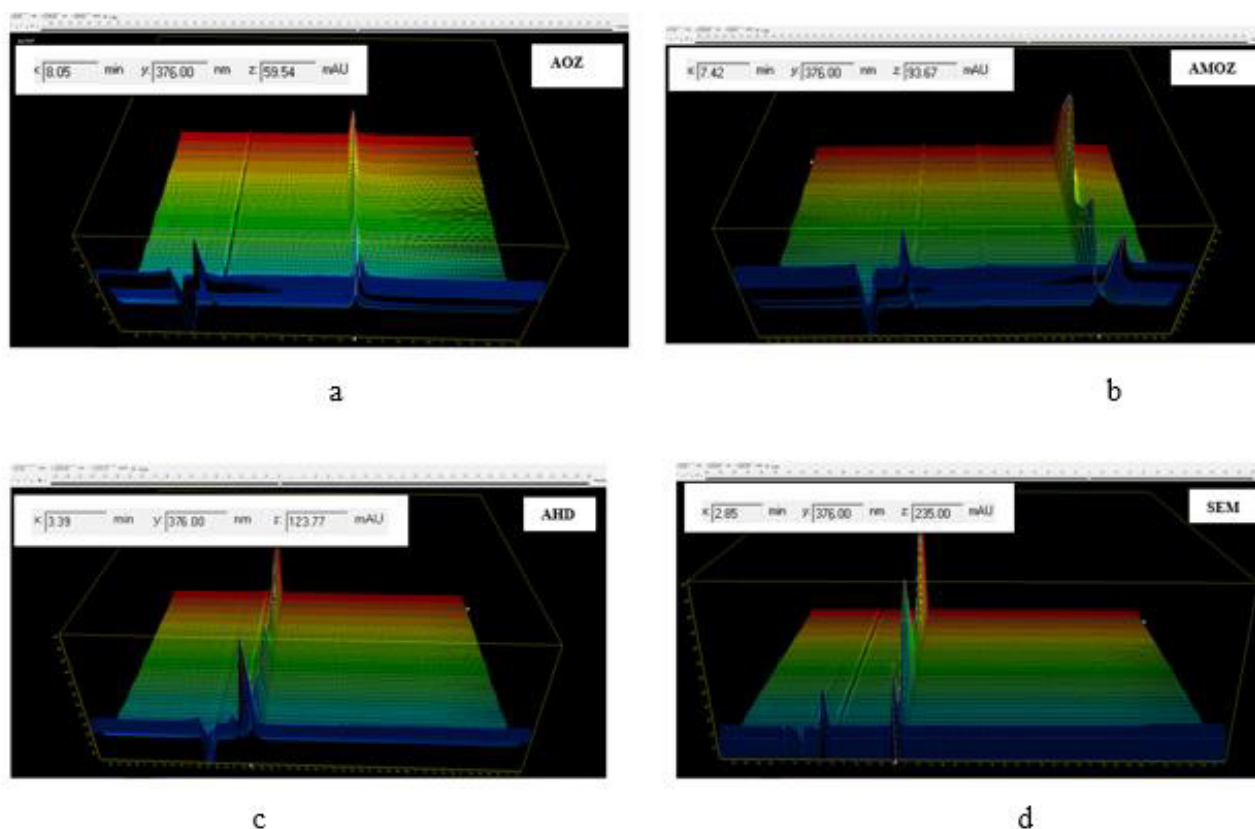


Figure 1: Scan results of nitrofurans metabolites in the range 190 - 600 nm a)– AOZ, b) AMOZ, c) AHD, d) SEM.

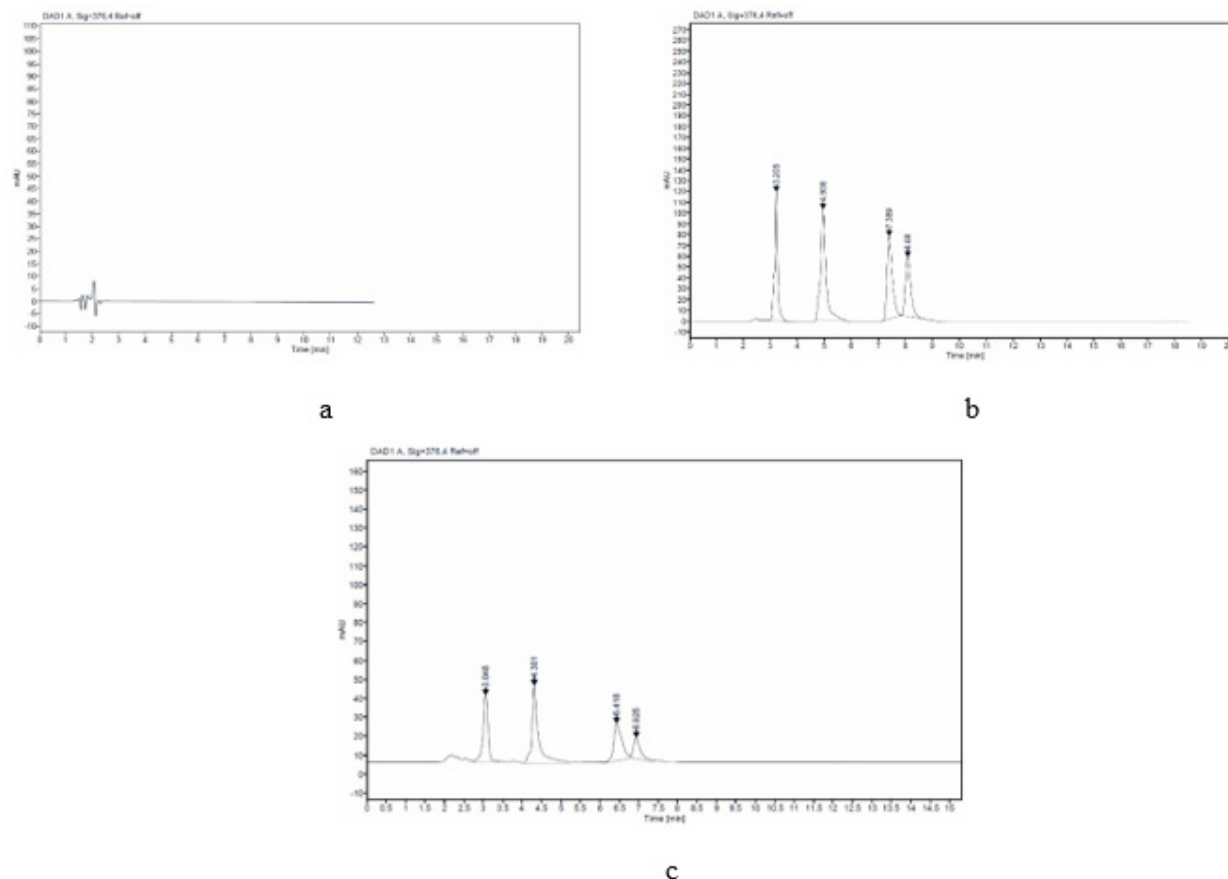


Figure 2: Chromatograms: a) Blank, b) Mix of standard samples of nitrofurans metabolites (AHD, SEM, AMOZ, AOZ); c) Metabolites of nitrofurans extracted from milk (AHD, SEM, AMOZ, AOZ).

LOD and LOQ

- For furazolidone (AOZ) S/N (signal to noise) at 0,32 $\mu\text{g}/\text{kg}$ is 3,2; at 0,91 $\mu\text{g}/\text{kg}$ 10,8.
- For furaltadone (AMOZ) S/N (signal to noise) at 0,35 $\mu\text{g}/\text{kg}$ is 2,8; at 0,95 $\mu\text{g}/\text{kg}$ - 9,9.

- For nitrofurantoin (AHD) S/N (signal to noise) at 0,29 $\mu\text{g}/\text{kg}$ is 3,0; at 0,91 $\mu\text{g}/\text{kg}$ - 10,5.
- For nitrofurazone (SEM) S/N (signal to noise) at 0,38 $\mu\text{g}/\text{kg}$ is 2,6; at 0,99 $\mu\text{g}/\text{kg}$ - 10,8.

The value of LOD and LOQ are presented in (Table 1).

Table 1: The mean value recovery, LOD, and LOQ of four compounds (n = 6).

Analytes	Mean Value of Recovery % (Standard 1 $\mu\text{g}/\text{kg}^{-1}$)	LOD	LOQ
AOZ	90,1	0,32	0,91
AMOZ	89,5	0,35	0,95
AHD	98,2	0,29	0,91
SEM	92,3	0,38	0,99

Linearity

To construct calibration graphs, we used 5 concentrations: 0.4, 0.8, 1, 1.4 and 1.8 $\mu\text{g}/\text{kg}$ for all four metabolites. The graphs are

linear in the indicated range and are acceptable as long as the correlation coefficient r^2 is above 0.999 (Figure 3).

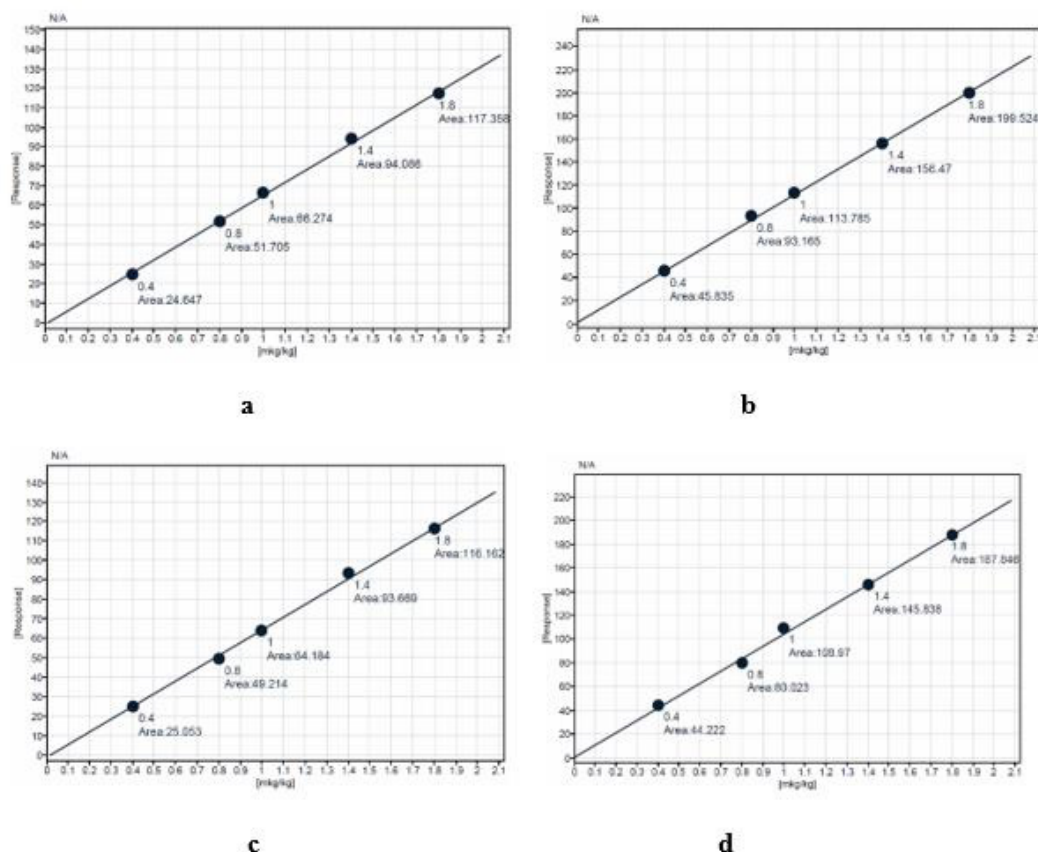


Figure 3: Calibration graphs of extracted nitrofurans metabolites: a)– AOZ, b) AMOZ, c) AHD, d) SEM.

Recovery

The method recoveries and RSDs were determined from 6 replicates at four concentration levels spiking blank samples over three days. The recovery results were observed in acceptable range

of 70-110%. All the data relating to method recovery and precision were summarised in (Table 4); mean recoveries ranging and CV% values were satisfactory, required by Decision 2002/657/EC [14]. The mean value of recovery, LOD, and LOQ were presented in (Table 2).

Table 2: The mean value $CC\alpha$ and $CC\beta$.

Analyte	Calibration Range (mg kg ⁻¹)	Linearity (r^2)	$CC\alpha$ (mg kg ⁻¹)	$CC\beta$ (mg kg ⁻¹)
AOZ	0,4 – 1,8	0,9995	0,24	0,27
AMOZ	0,4 – 1,8	0,9997	0,14	0,16
AHD	0,4 – 1,8	0,9992	0,18	0,22
SEM	0,4 – 1,8	0,9992	0,32	0,38

Decision limit ($CC\alpha$) and detection capability ($CC\beta$)

The $CC\alpha$ and $CC\beta$ for banned substances were calculated with the application of the following formula:

$$CC\alpha = C_1 + 2.33 \times SD_{wR}$$

where in C_1 is lowest concentration level of the validation study (MRPL) and SD_{wR} is the standard deviation from within-laboratory reproducibility.

$$CC\beta = CC\alpha + 1.64 \times SD_{wR,CC\alpha}$$

where in $SD_{wR,CC\alpha}$ is standard deviation at $CC\alpha$ concentration.

For each compound of $CC\alpha$ and $CC\beta$ were calculated from the linearity study.

The mean value $CC\alpha$ and $CC\beta$ were presented in (Table 2).

Discussion

Sample purification is important for increasing the sensitivity of the method, however; there are works in which analysis without purification is carried out. Fatih Alkan et al. nitrofurans metabolites

in honey, milk, poultry and fish samples were subjected to acid hydrolysis followed by derivatization with nitrobenzaldehyde and liquid-liquid extraction with ethyl acetate, and then without purification, LC-MS/MS determination [46]. In rare cases, liquid cleaning is used. An LC-ESI-MS-MS method for the analysis of metabolites of four nitrofurans (furazolidone, furaltadone, nitrofurazone and nitrofurantoin) in raw milk has been developed. The samples were achieved by hydrolysis of the protein-bound drug metabolites, derivatization with 2-nitrobenzaldehyde (2-NBA) and clean-up extraction liquid-liquid with ethyl acetate [47]; Bock et al. liquid-liquid separation is used to purify the obtained extracts [48,49]. Most often, solid-phase cleaning is used through a variety of expensive cartridges. Mottier et al. [50] quantitatively determined four nitrofurans in meat by isotope dilution liquid chromatography–electrospray ionisation–tandem mass spectrometry. This study used liquid–liquid extraction method and clean-up on a polymeric solid phase extraction cartridge (SPE) are then performed before LC–MS/MS analysis by positive electrospray ionisation (ESI) Barbosa et al. Determination of nitrofurans in animal feeds by liquid chromatography-UV photodiode array detection and liquid chromatography-ionspray tandem mass spectrometry. Following ethyl acetate extraction at mild alkaline conditions and purification on NH₂ column (SPE), the nitrofurans are determined using liquid chromatography with photodiode-array detection (LC-DAD) [51] Tomasz Śniegocki et al. After incubation the sample was purified by solid phase extraction technique [52] De La Calle et al. The sample was extracted with hydrochloric acid and derivatized with 2-nitrobenzaldehyde, with 1,2-[¹⁵N₂(¹³C)] SEM as the internal standard. The extract was neutralized and purified on a solid-phase extraction cartridge. In this study, a polystyrene–divinylbenzene copolymer (SDB–L) was used as sorbent material, solid phase extraction (SPE) with polymeric sorbent SEM was determined by reversed-phase LC with detection by MS/MS [53] The use of SPE technique was previously described by Leitner et al. [54] and they indicated that sample preparation protocol including cleanup with polymeric sorbent is simple and robust; Kaufmann et al. used an ultra-high performance liquid chromatography based method, coupled with high resolution mass spectrometry (UHPLC–HRMS), was developed to permit the detection and quantification of various nitrofurans and chloramphenicol residues in a number of animal based food products. This method is based on the hydrolysis of covalently bound metabolites and derivatization with 2-nitrobenzaldehyde. Clean-up is achieved by a liquid/liquid and a reversed phase/solid phase extraction [28] The method used by Pak-Sin et al. involves overnight acid hydrolysis and simultaneous derivatization of the released side chains with 2-nitrobenzaldehyde. During hydrolysis, the bound metabolites are hydrolyzed to the side chains. After pH adjustment and solid-phase extraction cleanup, the derivatives are detected and quantitated using a liquid chromatography–tandem mass spectrometry [55]. In accordance with the MRPL requirement We attempted to achieve the required detection limit by solid phase purification of the sample after extraction. For solid-phase cleaning, instead of expensive cartridges, we used columns with silica gel of our own preparation. As a result, we have obtained a sensitive, fast and relatively inexpensive method that meets the requirements of

MRPL. The method can be successfully used for the simultaneous determination of four metabolites of nitrofurans in milk and guarantees the safe use of milk and consequently products made from this milk in developing countries and regions.

Evaluation

This method has been developed and tested internally in accordance with the requirements of European Commission Decision 2002/657/EC [14]. The effectiveness of the method developed by us, is confirmed by the results of professional testing, implemented in the testing laboratory “GlobalTest” and accredited according to ISO 17025 by the Accreditation Agency of Georgia.

Conclusion

Developed methods indicate accordance with Decision 2002/657/EC [14,56-58]. The CC_α and CC_β are below the MRPL of 1 µg kg⁻¹. Due to its Cost effectiveness, it is available in developing countries and can be successfully used for the simultaneous determination of four metabolites of nitrofurans in milk, and guarantees the safe use of milk and, consequently, products made from this milk in developing countries and regions.

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Conflict of interests

The authors have no conflict of interests to declare.

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