



# Tailoring Phase Transition and Dielectric Response in BaTiO<sub>3</sub> Ceramics via Sb<sub>2</sub>O<sub>3</sub> Doping

Enxi Huang, Zhichen Pu, Meiyu Yang, Qixu Pan, Yuying Feng and Junjie Li\*

College of Optoelectronic Engineering, Chengdu University of Information Technology, Chengdu 610225, China

\*Corresponding author: Junjie Li, College of Optoelectronic Engineering, Chengdu University of Information Technology, Chengdu 610225, China.

Received Date: January 19, 2026

Published Date: February 19, 2026

## Abstract

The practical applications of ferroelectric materials are highly dependent on their phase transition behaviors and dielectric properties. To broaden the application scope of BaTiO<sub>3</sub>, direct Sb<sub>2</sub>O<sub>3</sub> introduction was adopted as a strategy to modulate the phase transition in this work. Experimental results demonstrate that Sb<sub>2</sub>O<sub>3</sub> doping induces a structural transformation of BaTiO<sub>3</sub> from tetragonal to pseudo-cubic symmetry, and facilitates the transformation of the original first-order phase transition to a diffuse type, while simultaneously shifting the Curie temperature toward room temperature. This modification pathway provides a promising avenue for optimizing the comprehensive performances of BaTiO<sub>3</sub> and expanding its application potential in pyroelectric, electrocaloric and energy storage fields.

## Introduction

Ferroelectric ceramics are extensively utilized in advanced fields such as pyroelectric sensors, energy storage devices, and electrocaloric refrigeration systems. These applications are highly reliant on their phase transition behavior and dielectric properties. For instance, remarkable pyroelectric and electrocaloric effects are typically achieved near the Curie temperature ( $T_c$ ), where the ferroelectric-to-paraelectric phase transition occurs [1]. Similarly, superior energy storage performance, characterized by low energy loss, requires excellent relaxor ferroelectric characteristics [2].

Barium titanate (BaTiO<sub>3</sub>) is a classic lead-free ferroelectric material that undergoes a sharp first-order ferroelectric-to-paraelectric phase transition around  $T_c \approx 125^\circ\text{C}$  [3]. This inherent feature severely restricts its room-temperature pyroelectric, electrocaloric performances, and energy storage capability. To broaden its application scope, cation substitution is the most conventional and effective strategy for tailoring the phase transition and dielectric response. Typical routes include replacing Ba<sup>2+</sup> with Ca<sup>2+</sup>, Sr<sup>2+</sup>, or rare-earth ions (RE<sup>3+</sup>, where RE = La, Ce, Nd, Sm, Eu, Gd, Dy, Er) and substituting Ti<sup>4+</sup> with Zr<sup>4+</sup>, Sn<sup>4+</sup>, or Hf<sup>4+</sup> [4].

Different from the conventional cation substitution strategy, this work proposes directly introducing the metal oxide Sb<sub>2</sub>O<sub>3</sub> into BaTiO<sub>3</sub> ceramics to tailor its phase transition and dielectric response. This novel modification approach is expected to optimize its pyroelectric, electrocaloric, and energy storage performances, thereby further expanding its application scope.

## Experimental

BaTiO<sub>3</sub>+xSb<sub>2</sub>O<sub>3</sub> ceramics (x = 0, 0.01, 0.02, 0.03) were prepared by a conventional solid-state reaction method, using BaCO<sub>3</sub>, TiO<sub>2</sub>, Sb<sub>2</sub>O<sub>3</sub>. The homogenized mixed powders were calcinated at 1150°C for 3 h, followed by sintering at 1350°C for 2 h to obtain bulk ceramics. The phase structures were characterized by an X-ray diffractometer (XRD, Shimadzu LabX, Japan). A differential scanning calorimeter (DSC, TA Q2000, USA) was employed to collect heat flow data. Dielectric properties were measured by precision impedance analyzer (Wayne Kerr WK6500P, UK) coupled with a temperature-control and data-collect system (BALAB DMS-2000, China).

## Results and Discussion

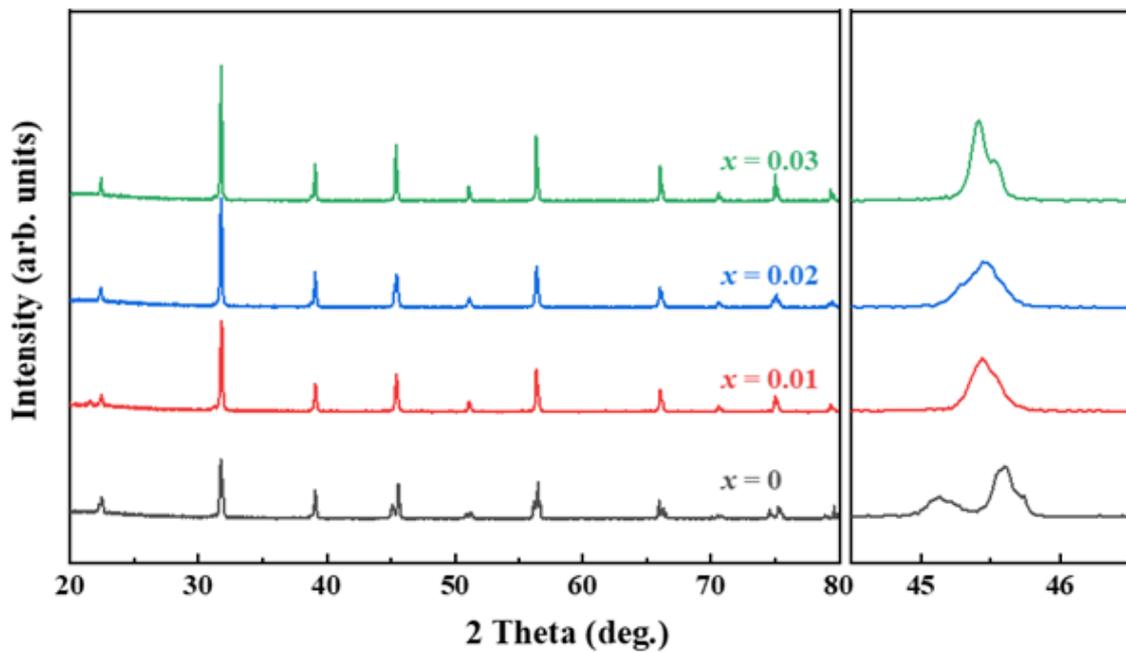


Figure 1: XRD patterns of  $\text{BaTiO}_3+x\text{Sb}_2\text{O}_3$  ceramics.

Figure 1 presents the XRD patterns of  $\text{BaTiO}_3+x\text{Sb}_2\text{O}_3$  ceramics. All samples exhibit a pure perovskite structure without detectable secondary phases, indicating that  $\text{Sb}_2\text{O}_3$  has been fully incorporated into the  $\text{BaTiO}_3$  lattice. Undoped  $\text{BaTiO}_3$  displays a distinct (002)/(200) peak splitting at  $2\theta \approx 45^\circ$ , which is consistent with tetragonal

(T)  $P4mm$  symmetry [5]. In contrast, a single diffraction peak is obtained near  $2\theta \approx 45^\circ$  for doped  $\text{BaTiO}_3$ , manifesting a pseudo-cubic structure. This result demonstrates that the introduction of  $\text{Sb}_2\text{O}_3$  disturbs the long-range ordered lattice structure of  $\text{BaTiO}_3$ .

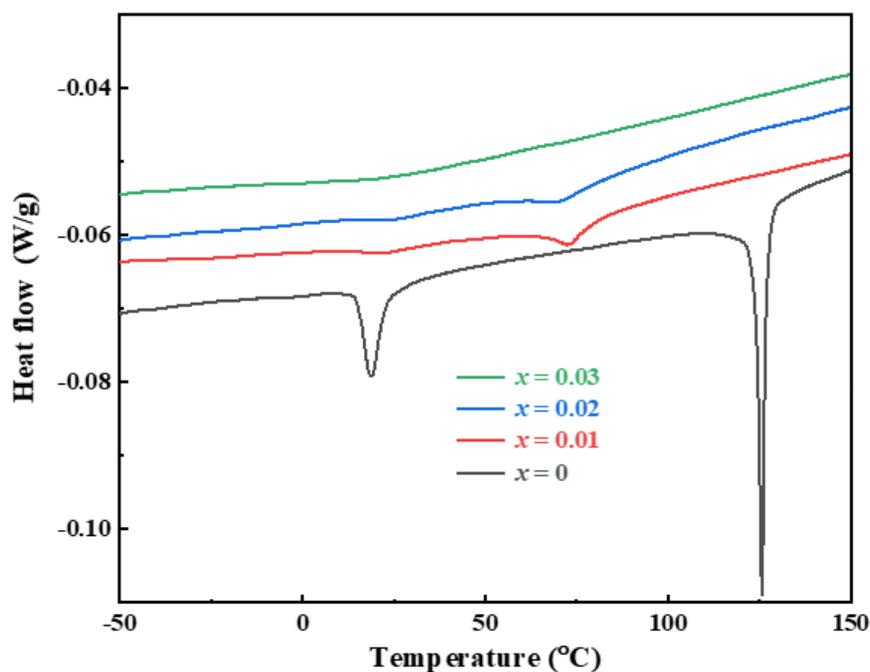
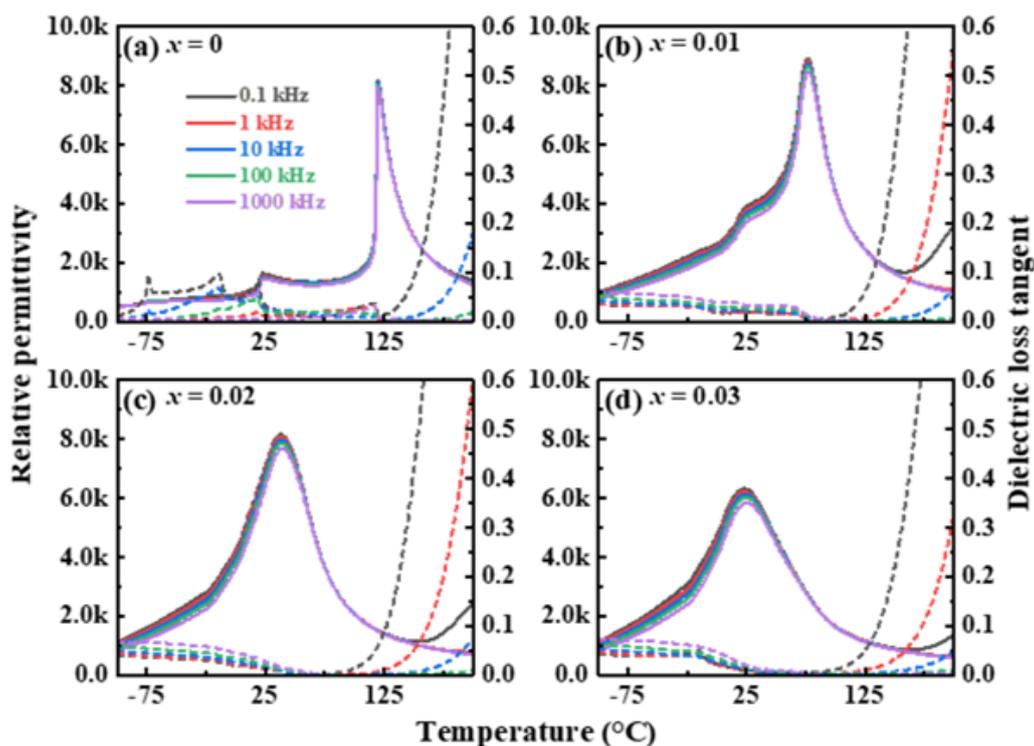


Figure 2: Heat flow curves of  $\text{BaTiO}_3+x\text{Sb}_2\text{O}_3$  ceramics.

Figure 2 presents the heat flow curves of  $\text{BaTiO}_3+x\text{Sb}_2\text{O}_3$  ceramics measured in the temperature range of  $-50^\circ\text{C}$  to  $150^\circ\text{C}$  during heating. For pure  $\text{BaTiO}_3$ , two sharp endothermic peaks are observed, among which, the low-temperature peak near  $19^\circ\text{C}$  corresponds to orthorhombic-tetragonal (O-T) first-order ferroelectric phase transition, while the high-temperature peak at  $125^\circ\text{C}$ , i.e.  $T_c$ , arises from the tetragonal-cubic (T-C) first-order

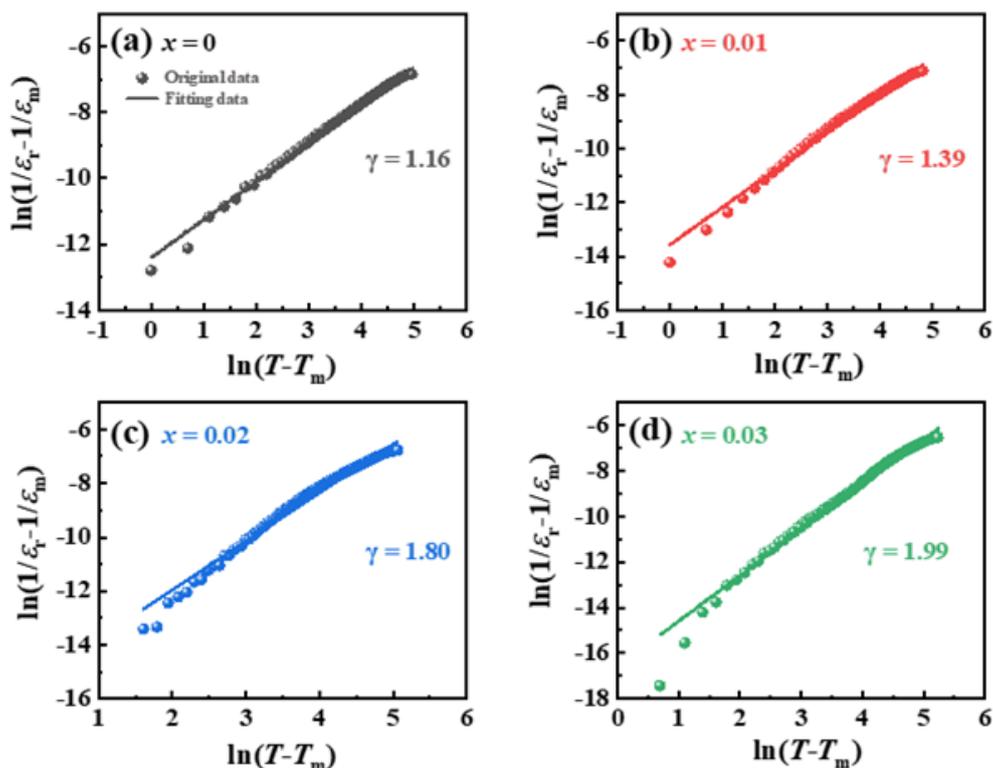
ferroelectric-to-paraelectric phase transition [6]. When  $\text{Sb}_2\text{O}_3$  is doped at  $x=0.01$ , both endothermic peaks weaken significantly, and TC shifts to a lower temperature ( $75^\circ\text{C}$ ). With further increasing  $\text{Sb}_2\text{O}_3$  doping content, the endothermic peaks gradually diminish. This indicates that  $\text{Sb}_2\text{O}_3$  doping facilitates the transformation of the phase transition from a first-order to a diffuse type.



**Figure 3:** Temperature dependences of dielectric constants and losses of  $\text{BaTiO}_3+x\text{Sb}_2\text{O}_3$  ceramics.

Figure 3 illustrates the temperature dependences of dielectric constants ( $\epsilon_r$ ) and losses of  $\text{BaTiO}_3+x\text{Sb}_2\text{O}_3$  ceramics at frequencies ranging from 0.1 to 1000 kHz. Consistent with the heat flow results, pure  $\text{BaTiO}_3$  exhibits first-order phase transitions around  $19^\circ\text{C}$  (O-T) and  $125^\circ\text{C}$  (T-C). Additionally, an extra dielectric peak is observed at  $-75^\circ\text{C}$ , corresponding to the rhombohedral-orthorhombic (R-O) phase transition [6]. For the  $x=0.01$  doped sample, the  $T_c$  shifts to  $77^\circ\text{C}$ , and the dielectric peaks corresponding to the R-O and O-T phase transitions become flattened. Notably,  $x=0.02$  and  $x=0.03$  samples only show a single dielectric peak corresponding to  $T_c$  at  $38^\circ\text{C}$  and  $22^\circ\text{C}$ , respectively, which is conducive to their room-temperature applications.

To further quantify the phase transition diffuseness, the modified Curie-Weiss law was employed [7]. Figure 4 displays the plots and linear fittings of  $\ln(1/\epsilon_r - 1/\epsilon_m)$  and  $\ln(T-T_m)$  at 1000 kHz, where  $\epsilon_m$  is the maximum dielectric constant and  $T_m$  is the temperature corresponding to  $\epsilon_m$ . All sample data can be well fitted linearly, and the slope of the fitting line is defined as the diffusion factor ( $\gamma$ ). With the increase of  $\text{Sb}_2\text{O}_3$  doping content from  $x=0$  to  $x=0.03$ , the diffusive factor increases from  $\gamma=1.05$  to  $\gamma=1.99$ . A  $\gamma$  value close to 1 indicates a normal ferroelectric, while a  $\gamma$  value approaching 2 is characteristic of a relaxor ferroelectric. This result further confirms that  $\text{Sb}_2\text{O}_3$  doping converts the phase transition of  $\text{BaTiO}_3$  from a first-order to a diffuse type.



**Figure 4:** The plots and linear fittings of  $\ln(1/\epsilon_r - 1/\epsilon_m)$  vs.  $\ln(T - T_m)$  at 1000 kHz.

## Conclusion

This work achieves  $\text{BaTiO}_3$  modification via direct  $\text{Sb}_2\text{O}_3$  introduction. The results show that  $\text{Sb}_2\text{O}_3$  can be fully dissolved into the  $\text{BaTiO}_3$  lattice to form a pure perovskite phase, triggering a structural transformation from tetragonal to pseudo-cubic symmetry. Meanwhile, it converts original first-order phase transition of  $\text{BaTiO}_3$  to a diffuse type, endowing the material with relaxor ferroelectric properties, and shifts the Curie temperature toward room temperature. This novel modification strategy provides effective support for optimizing the performances of  $\text{BaTiO}_3$  and expanding its application prospects.

## Acknowledgement

This work was supported by the National Undergraduate Innovation and Entrepreneurship Training Program (No. 202410621019).

## References

1. Y Hou, J Li, R Yin, X Su, Y Su, et al. (2025) The critical role of phase transition and composition regulation in inorganic perovskite electrocaloric materials. *J Mater Chem C* 13: 5406-5423.
2. L Yang, X Kong, F Li, H Hao, Z Cheng, et al. (2019) Perovskite lead-free dielectrics for energy storage applications. *Prog Mater Sci* 102: 72-108.
3. X Moya, E Stern-Taulats, S Crossley, D González-Alonso, S Kar-Narayan, et al. (2013) Giant Electrocaloric Strength in Single-Crystal  $\text{BaTiO}_3$ . *Adv Mater* 25(9): 1360-1365.
4. C Zhao, Y Huang, J Wu (2020) Multifunctional barium titanate ceramics via chemical modification tuning phase structure. *InfoMat* 2: 1163-1190.
5. G Canu, G Confalonieri, M Deluca, L Curecheriu, M T Buscaglia, et al. (2018) Structure-property correlations and origin of relaxor behaviour in  $\text{BaCexTi}_{1-x}\text{O}_3$ . *Acta Mater* 152: 258-268.
6. J Li, D Zhang, S Qin, T Li, M Wu, et al. (2016) Large room-temperature electrocaloric effect in lead-free  $\text{BaHf}_x\text{Ti}_{1-x}\text{O}_3$  ceramics under low electric field. *Acta Mater* 115: 58-67.
7. J Li, R Yin, Z Xiong, Y Bao, X Zhang, et al. (2024) Manipulating Zr/Ti ratio based on phase diagram for large electrocaloric effects with multiple target operation temperatures in PLZT ceramics. *J Adv Ceram* 13: 1422-1431.